

A NEW TRITERPENE FROM THE LEAVES OF *Betula costata*

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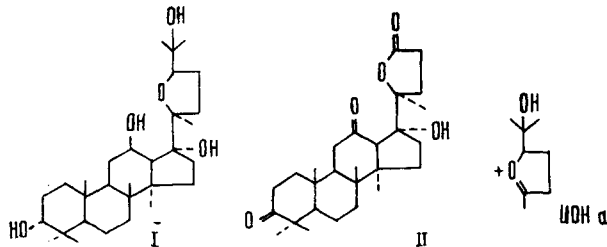
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From the unsaponifiable part of an ethereal extract of the leaves of *Betula costata* (collected in the Maritime Territory, Shkotovo region, at the beginning of July by P. G. Gorov) by separation on a chromatographic column containing silica gel (KSK) in the benzene-ethyl acetate system we have isolated a triterpene (I) with the composition  $C_{30}H_{52}O_5$  (yield on the weight of the air-dry leaves 0.12%) with mp 250-251.5°C (acetone),  $[\alpha]_D^{20} + 6^\circ$  (c 0.5; chloroform). The IR spectrum of (I) ( $CHCl_3$ ) shows the bands of hydroxyl absorption at  $3420\text{ cm}^{-1}$  (not depending on the concentration) and  $3610\text{ cm}^{-1}$ .

The ratio of the intensities of the peaks with m/e 477 ( $M - CH_3$ ) and 474 ( $M - H_2O$ ) in the mass spectrum of (I) and the considerable peaks with m/e 349 (tetracyclic system) and 143 (side chain) showed that substance (I) is a tetracyclic triterpene with a side chain in the form of a substituted tetrahydrofuran ring. According to the results of mass spectroscopy of the oxide of betulafolientriol [1], the peak with m/e 143 in the mass spectrum of (I) must belong to the ion  $\alpha$  showing the presence of a side chain of the same type as in betulafolientriol oxide.

In the NMR spectrum ( $\nu = 90\text{ MHz}$ , in  $CDCl_3$ ) of the triterpene (I) the signals of the protons of eight methyl groups were found at ( $\delta$ , ppm): 0.83 (3 H, s), 0.86 (3 H, s), 0.93 (3 H, s), 0.97 (3 H, s), 1.11 (3 H, s), 1.18 (3 H, s), 1.29 (3 H, s), and 1.34 (3 H, s) and the signals of the protons on the corresponding carbon atoms at: 3.39 (1 H, triplet,  $J = 2.2\text{ Hz}$ ; proton at  $C_3$ ), 3.80 ppm (1 H, sextet,  $J = 11\text{ Hz}$  and  $J = 7\text{ Hz}$ ); proton at  $C_{12}$ ; 3.74 (1 H, triplet,  $J = 6.6\text{ Hz}$ ; proton at  $C_{24}$ ); and 5.24 (1 H, broad singlet), corresponding to the proton of the hydroxy group at  $C_{12}$  bound by an intramolecular hydrogen bond with the oxygen of the tetrahydrofuran ring.

When compound (I) was oxidized with the Jones reagent in acetone, by analogy with [1], the hydroxy diketo lactone (II) was obtained which had similar characteristics to the hydroxy diketo lactone obtained by Fischer and Seiler from betulafolientetraol [2] with mp 280-290°C (decomp.),  $[\alpha]_D^{20} + 49.0^\circ$  (c 1.0; chloroform). The IR spectrum of (II) (KBr) showed absorption in the  $1770\text{-cm}^{-1}$  region ( $\gamma$ -lactone) and at  $1750\text{ cm}^{-1}$  (CO) in  $CHCl_3$ , there was absorption in the  $3450\text{-cm}^{-1}$  region (OH). In the NMR spectrum of (II) there was a signal at  $\delta 5.31\text{ ppm}$  (1 H, s) corresponding to the proton of the hydroxy group at  $C_{17}$  bound by an intramolecular hydrogen bond with the carbonyl at  $C_{12}$ .



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On the basis of these results, structure (I) is suggested as the most probable for this new triterpene.

LITERATURE CITED

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2. F. G. Fischer and N. Sieler, Ann. Chem., 644, 146 (1961).